Amalgamation and depiction of ZnO nano particles by Gigantic-Swallow-Wort plants with in immature compound diminution technique

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ABSTRACT : In attendance revise focuses on the extraction of ZnO nano particles by immature chemical uction technique from the bio components of leaves extract of Gigantic Swallow Wort. X-Ray Diffraction (XRD), FT-IR Spectroscopy characterizations was done for synthesized ZnO nanoparticles. X- ray diffraction studies showed that the particles are hexagonal in nature **Keywords**— FTIR, Green chemiccal reduction mtehod. XRD,

1. INTRODUCTION

Semiconductors with dimensions in the nanometer realm are important because their electrical, ical and chemical properties can be tuned by changing the size of particles. But Zinc oxide (ZnO) is a wide band gap (3.37 eV) and high excitation binding energy of (60 meV) at room temperature [1,2] and has unique optical and as well as excellent thermal and chemical stability [3]. ZnO nanoparticles have very large surface area with potentially low manufacturing cost. The shape and size of nano materials immensely affect their physical and chemical properties. Among the metal oxide nanoparticles, zinc oxide is interesting because it has vast applications in various areas such as optical, piezoelectric, magnetic, and gas sensing. Besides these properties, ZnO nanostructure exhibits high catalytic efficiency, strong adsorption ability and are used more and more frequently in the manufacture of sunscreens [4], ceramics and rubber processing, wastewater treatment, and as a fungicide [5, 6]. In fact, nZnO usage may overtake nano-titanium dioxide (nTiO2) in the near future as it can absorb both UV-A and UV-B radiation while nTiO2 can only block UV-B, and thereby offering better protection and improved opaqueness [5]. Several physical and chemical procedures have been used for the synthesis of large quantities of metal nanoparticles in relatively short period of time. Chemical methods lead to the presence of some toxic chemicals adsorbed on the surface that may have adverse effects in medical application [7]. Currently, plant-mediated biological synthesis of nanoparticles is gaining importance due to its simplicity, eco-friendliness and extensive antimicrobial activity [8, 9]. It has wide applications in different industries including photodetectors[10], sensors [11], solar cells [12], antibacterial or medical products [13-15], cosmetics [16], etc. ZnO nanoparticles can be synthesized by various chemical for physical methods such as precipitation [17], solgel [18], solvo/hydrothermal [19], chemical vapor deposition [20], spray pyrolysis [21], etc. Biosynthesis of zinc oxide nanoparticles by plants such as Aloe vera [22] and gold nanoparticles by alfalfa [23,24], Cinnamomum camphora[25], neem [26], Emblica officianalis [27], lemongrass [28] and tamarind [29] have been reported. To the best of our knowledge, the present study is the first report on ZnO nanoparticles synthesized using Gigantic Swallow Wort leaves

2. MATERIALS AND METHODS

A. Materials

Zinc nitrate and utilized ingredients with analytical grade chemicals were purchased from Merck and used without further purification. Distilled and deionized water was used in this work. The leaves of Gigantic Swallow Wort (Figure 1) plant collected from our surround places Zinc nitrate is a crystalline and inorganic chemical compound with the formula Zn(NO3)2. Zinc Nitrate is soluble in both alcohol and water. The structure of zinc nitrate is shown in figure 1.



Figure 1 structure of zinc nitrate

B. Preparation of leaf extract

For the preparation of leaves extract fresh leaves were collected from Gigantic Swallow Wort plants (figure 2). The leaves were washed several times with water to remove the dust particles and then sun dried to remove the residual moisture.



Figure 2 Gigantic Swallow Wort plant

The dried leaves were cutted and grinded for powder. Then taking 20 gm of dried Gigantic Swallow Wort leaves boiled in 250 ml of deionised water for one hour at 70 to 80 degree centegrdes. The mixture solution cooled at room temperature. The leaves extract (Figure 3) filtered by using watts man filter paper.

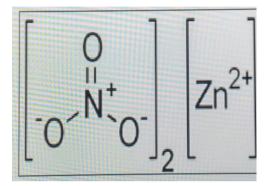


Figure 3. Gigantic Swallow Wort plant leaf extract

C. Green synthesis of ZnO nanoparticles using leaf Extract of Giganticon Swallow Wort Plant.

For the ZnO nanoparticles synthesis, 100 ml of Gigantic Swallow Wort leaf extract was taken boiled to 60-70 C. using magnetic stirrer heater. 5 grams of Zinc Nitrate was added to the solution. This mixture is then boiled until it reduced to a deep yellow coloured paste. This paste was then collected in a ceramic crucible and heated in an air heated furnace at 300 degree Celsius for 2 hours. A light yellow coloured powder (figure 3) was obtained and this was carefully collected and packed for characterization purposes.

A. FTIR Spectra analysis

3. RESULTS AND DISCSSION

The FTIR spectrum of ZnO nanoparticles is shown in Figure 4. The IR spectrum of transmittance was taken by using a Bruker FT –IR instrument operating at a resolution of 2000-400 cm-1. In IR spectra, the absorption peak at 473.33 cm⁻¹ indicates the presence of ZnO nanoparticles. The sharp characteristic peaks are also observed in FTIR spectrum of ZnO nanoparticles synthesized from Zinc Nitrate suggesting the high crystalline nature of ZnO nanoparticles.

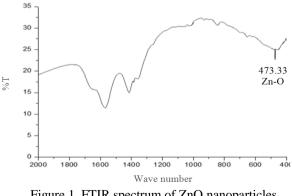


Figure 1. FTIR spectrum of ZnO nanoparticles B. X- ray diffraction analysis of ZnO nanoparticles.

XRD spectrum of the prepared ZnO nanoparticles was carried out using XRD for 2θ values ranging from 20 to 70° using CuK α radiation at $\lambda = 1.5406$ Å. In ZnO, the 2θ values with (hkl) plane at 33.5°(100), 35.2°(002), 37.1°(101), 47.4°(102), 57.5°(110), 62.8°(103) and 67.8°(201) were observed. The spectrum (Figure 5.) confirmed the hexagonal zinc oxide structure for ZnO nanoparticles. The average particle size (D) of synthesized nanoparticles was calculated using the well known Scherrer formula

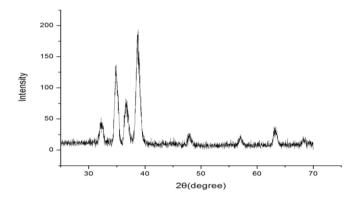


Figure 2 X-Ray diffraction spectrum of ZnO nanoparticles

4. CONCLUSION

ZnO NPs was synthesized by the green chemical reduction method using Gigantic-Swallow-Wort leaf extract is simple and cost effective. The as prepared ZnO nanoparticles were characterized using several techniques such as XRD and FTIR. The FT-IR studies showed an absorption peak at 473.33 cm⁻¹ (Zn-O linkage) which indicated the formation of zinc oxide nanoparticles. From XRD analysis, the structure of the ZnO particles is confirmed as hexagonal with average particle size 28.38 nm.

REFERENCES

- [1]. E. Roduner, Size matters: why nanomaterials are different. Chemical Society Reviews 35(7), (2006), 583-592.
- [2]. M.R Jones et al., Templated Techniques for the Synthesis and Assembly of Plasmonic Nanostructures., Chemical Reviews, 111(6), (2011), 3736-3827.
- [3]. S.Iravani, Green synthesis of metal nanoparticles using plants. Green Chemistry. 13(10), (2013), 2638-2650
- [4]. R. Seshadri, in: C.N.R. Rao, A. Muller, A.K. Cheetham (Eds.), "The Chemistry of Nanomaterials, Vol.1, Wiley-VCH Verlag GmbH, Weinheim, 2004, pp. 94–112.
- [5]. L. Theodore, Nanotechnology: Basic Calculations for Engineers and Scientists, Wiley, Hoboken, 2006.
- [6]. X. Wang, J. Lu, M. Xu, B. Xing, Sorption of pyrene by regular and nanoscaled metal oxide particles: influence of adsorbed organic matter, Environmental Science and Technology 42 (2008) 7267–7272.
- [7]. D. Jain, H.K. Daima, S. Kachhwaha, S.L. Kothari, Synthesis of plant-mediated silver nanoparticles using papaya fruit extract and evaluation of their antimicrobial activities, Digest Journal of Nanomaterials and Biostructures 4 (2009) 557–563.
- [8]. A. Saxena, R.M. Tripathi, R.P. Singh, Biological synthesis of silver nanoparticles by using onion (Allium cepa) extract and their antibacterial activity, Digest Journal of Nanomaterials and Biostructures 5 (2010) 427–432.
- [9]. N. Khandelwal, A. Singh, D. Jain, M.K. Upadhyay, H.N. Verma, Green synthesis of silver nanoparticles using Argimone mexicana leaf extract and evaluation of their antimicrobial activities, Digest Journal of Nanomaterials and Biostructures 5 (2010) 483–489.
- [10]. Liang S, Sheng H, Liu Y, Huo Z, Lu Y, Shen H: "ZnO Schottky ultraviolet photodetectors". J Cryst Growth 2001, 225(no. 2–4):110–113.
- [11]. Lin Y, Zhang Z, Tang Z, Yuan F, Li J: Characterisation of ZnO-based Varistors Prepared from Nanometre Precursor Powders. Adv. Mater. Opt. Electron 1999, 9:205–209.

- [12]. Golego N, Studenikin SA, Cocivera M: "Sensor Photoresponse of Thin-Film Oxides of Zinc and Titanium to Oxygen Gas". J Electrochem Soc 2000, 147:1592.
- [13]. Sawai J: Quantitative evaluation of antibacterial activities of metallic oxide powders (ZnO, MgO and CaO) by conductimetric assay. J Microbiol Methods 2003, 54(2):177–182.
- [14]. Zhang L, Jiang Y, Ding Y, Povey M, York D: Investigation into the antibacterial behaviour of suspensions of ZnO nanoparticles (ZnO nanofluids). Journal of Nanoparticle Research 2006, 9(3):479–489.
- [15]. Raghupathi KR, Koodali RT, Manna AC: Size-dependent bacterial growth inhibition and mechanism of antibacterial activity of zinc oxide nanoparticles. ACS 2011, 27(7):4020–4028.
- [16]. Vaezi MR, Sadrnezhaad SK: Nanopowder synthesis of zinc oxide via solochemical processing. Mat Des 2007, 28(2):515–519.
- [17]. Wang L, Muhammed M: Synthesis of zinc oxide nanoparticles with controlled morphology". J Mater Chem 1999, 9:2871–2878.
- [18]. Yan C, Chen Z, Zhao X: Enhanced electroluminescence of ZnO nanocrystalline annealing from mesoporous precursors. Solid State Commun 2006, 140(1):18–22.
- [19]. Pan A, Yu R, Xie S, Zhang Z, Jin C, Zou B: ZnO flowers made up of thin nanosheets and their optical properties. J Cryst Growth 2005, 282(1–2):165–172.
- [20]. Wu J-J, Liu S-C: Catalyst-Free Growth and Characterization of ZnO Nanorods. J Phys Chem B 2002, 106(37):9546–9551.
- [21]. Ghaffarian H, Saiedi M: "Synthesis of ZnO Nanoparticles by Spray Pyrolysis Method". Iran J Chem Chem Eng 2011, 30(no. 1):1–6.
- [22]. G. Sangeetha, S. Rajeshwari, R. Venckatesh, Green synthesis of zinc oxide nanoparticles by aloe barbadensis miller leaf extract:structure and optical properties, Materials Research Bulletin 46 (2011) 2560–2566.
- [23]. J.L. Gardea-Torresdey, J.G. Parsons, E. Gomez, J. Peralt Videa, Formation and growth of Au nanoparticles inside live alfalfa plants, Nanoletters 2 (2002) 397–401.
- [24]. J.L. Gardea-Torresdey, E. Gomez Jr., Peralta-VideaJr, J.G. Parsons, H. TroianiM. Jose-Yacaman, Alfalfa Sprouts: A Natural Source for the Synthesis of Silver Nanoparticles, Langmuir 19 (2003) 1357–1361.
- [25]. J. Huang, Q. Li, D. Sun, Y. Lu, Y. Su, X. Yang, Biosynthesis of silver and gold nanoparticles by novel sundried Cinnamomum camphora leaf, Nanotechnology 18 (2007) 105104–105114.
- [26]. S. Shiv Shankar, A. Rai, A. Ahmad, M. Sastry, Rapid synthesis of Au, Ag, and bimetallic Au core–Ag shell nanoparticles using neem (Azadirachta indica) leaf broth, Journal of Colloid and Interface Science 275 (2004) 496–502.
- [27]. B. Ankamwar, D. Chinmay, A. Absar, S. Murali, Biosynthesis of gold and silver nanoparticles using emblica officinalis fruit extract, their phase transfer and transmetallation in an organic solution, Journal of Nanoscience and Nanotechnology 10 (2005) 1665–1671.
- [28]. S. Shiv Shankar, A. Rai, A. Ahmad, M. Sastry, Rapid synthesis of Au, Ag, and bimetallic Au core–Ag shell nanoparticles using neem (Azadirachta indica) leaf broth, Journal of Colloid and Interface Science 275 (2004) 496–502.
- [29]. B. Ankamwar, M. Chaudhary, Gold nanotriangles biologically synthesized using tamarind leaf extract and potential application in vapor sensing, Synthesis and Reactivity in Inorganic and Metal–Organic Chemistry 35 (2005) 19–26.